# organic papers

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#### Key indicators

Single-crystal X-ray study T = 150 KMean  $\sigma$ (C–C) = 0.002 Å R factor = 0.053 wR factor = 0.135 Data-to-parameter ratio = 19.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 3,4-Lutidine

The crystal structure of 3,4-lutidine (3,4-dimethylpyridine,  $C_7H_9N$ ), has been determined at 150 (2) K following *in situ* crystal growth from the liquid. In space group C2/c, there are four independent molecules in the asymmetric unit, linked into dimers *via*  $C-H \cdots N$  interactions.

#### Comment

As part of a study devoted to improving the techniques for determining the crystal structures of substances that are liquids at room temperature, we have reported previously the crystal structures of 2,6-lutidine (Bond *et al.*, 2001), 3,5-lutidine (Bond & Davies, 2002*a*) and 2,5-lutidine (Bond & Davies, 2002*b*). We report here the crystal structure of 3,4-lutidine, (I), determined at 150 (2) K following *in situ* crystal growth from the liquid.



Compound (I) crystallizes in the monoclinic space group C2/c with four independent molecules in the asymmetric unit (Fig. 1). Within the asymmetric unit, molecules are linked into dimers via  $C-H \cdot \cdot \cdot N$  interactions (Table 1). In the structures of the other lutidines reported to date, molecules are linked into linear chains via C-H···N interactions involving the H atom at the 4-position; this interaction is clearly prohibited in (I). Two orientations may be envisaged for dimerization in which a centrosymmetric motif would result: either both molecules interact through the H atoms at their 2-positions, or both interact through the H atoms at their 6-positions. The observed dimer involves one molecule of (I) interacting through H at the 2-position and one interacting through H at the 6-position, giving rise to an asymmetric motif (point symmetry 1). In both independent dimers within the asymmetric unit, the interaction C6-H6...N1 is significantly shorter and closer to linear than the C2-H2···N1 interaction (Table 1). This is not obviously an intra-dimer effect and may be a result of inter-dimer interactions between methyl substituents. The dimers may be considered to form layers parallel to (001), with the planes through the dimer units lying alternately parallel to (110) and ( $\overline{110}$ ) in adjacent layers (Figs. 2 and 3).

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## Figure 1

The asymmetric unit and atom-labelling scheme in (I), showing displacement ellipsoids (C/N atoms) at the 50% probability level (XP; Sheldrick, 1993). Independent molecules are denoted by the suffixes A, B, C and D.



## Figure 3

Projection on to (001) of the entire structure of (I), showing layers of dimers oriented alternately parallel to (110) and ( $\overline{110}$ ) (CAMERON; Watkin et al., 1996).

# **Experimental**

The sample (99%) was obtained from the Aldrich company and used without further purification. The crystal was grown in a 0.3 mm glass capillary tube at ca 235 K (a temperature only slightly less than the melting point of the solid in the capillary) using a technique described earlier (Davies & Bond, 2001). Once grown, the crystal was cooled to 150 (2) K for data collection. The length of the cylindrical crystal was not estimated, but it exceeded the diameter of the collimator (0.35 mm).



#### Figure 2

Projection on to (001) of a single layer in (I), showing dimers linked by C-H···N interactions orientated parallel to (110) (CAMERON; Watkin et al., 1996).

#### Crystal data

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C <sub>7</sub> H <sub>9</sub> N	$D_x = 1.124 \text{ Mg m}^{-3}$
$M_r = 107.15$	Mo $K\alpha$ radiation
Monoclinic, C2/c	Cell parameters from 9883
a = 20.9006 (8) Å	reflections
b = 20.6112 (9) Å	$\theta = 1.0-27.5^{\circ}$
c = 12.9921 (3) Å	$\mu = 0.07 \text{ mm}^{-1}$
$\beta = 115.135 \ (2)^{\circ}$	T = 150 (2)  K
$V = 5066.9 (3) \text{ Å}^3$	Cylinder, colourless
Z = 32	0.15 mm (radius)

#### Data collection

Nonius KappaCCD diffractometer	$R_{\rm int} = 0.026$
Thin-slice $\omega$ and $\varphi$ scans	$\theta_{\rm max} = 27.5^{\circ}$
Absorption correction: none	$h = 0 \rightarrow 26$
752 measured reflections	$k = 0 \rightarrow 26$
740 independent reflections	$l = -16 \rightarrow 15$
041 reflections with $I > 2\sigma(I)$	

#### Refinement

Refinement on  $F^2$ 
$$\begin{split} R[F^2 > 2\sigma(F^2)] &= 0.053 \\ wR(F^2) &= 0.135 \end{split}$$
S = 1.035740 reflections 301 parameters H-atom parameters constrained

## Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C6A - H6A \cdots N1B$	0.95	2.48	3.368 (2)	156
$C2B - H2B \cdot \cdot \cdot N1A$	0.95	2.83	3.630 (2)	143
$C6C - H6C \cdot \cdot \cdot N1D$	0.95	2.45	3.339 (2)	156
$C2D - H2D \cdots N1C$	0.95	2.88	3.661 (2)	140

H atoms were placed geometrically and refined with isotropic displacement parameters, with common parameters assigned to

 $w = 1/[\sigma^2(F_o^2) + (0.0488P)^2]$ 

where  $P = (F_o^2 + 2F_c^2)/3$ 

+ 2.9233P

 $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta \rho_{\rm max} = 0.20 \text{ e } \text{\AA}^{-3}$ 

 $\Delta \rho_{\rm min} = -0.16 \ {\rm e} \ {\rm \AA}^{-3}$ 

chemically equivalent H atoms (one parameter for all methyl H atoms, four parameters in total). Each methyl group was allowed to rotate about its local threefold axis.

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *HKL DENZO* and *SCALEPACK* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SIR*-92 (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *XP* (Sheldrick, 1993) and *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *SHELXL*97.

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